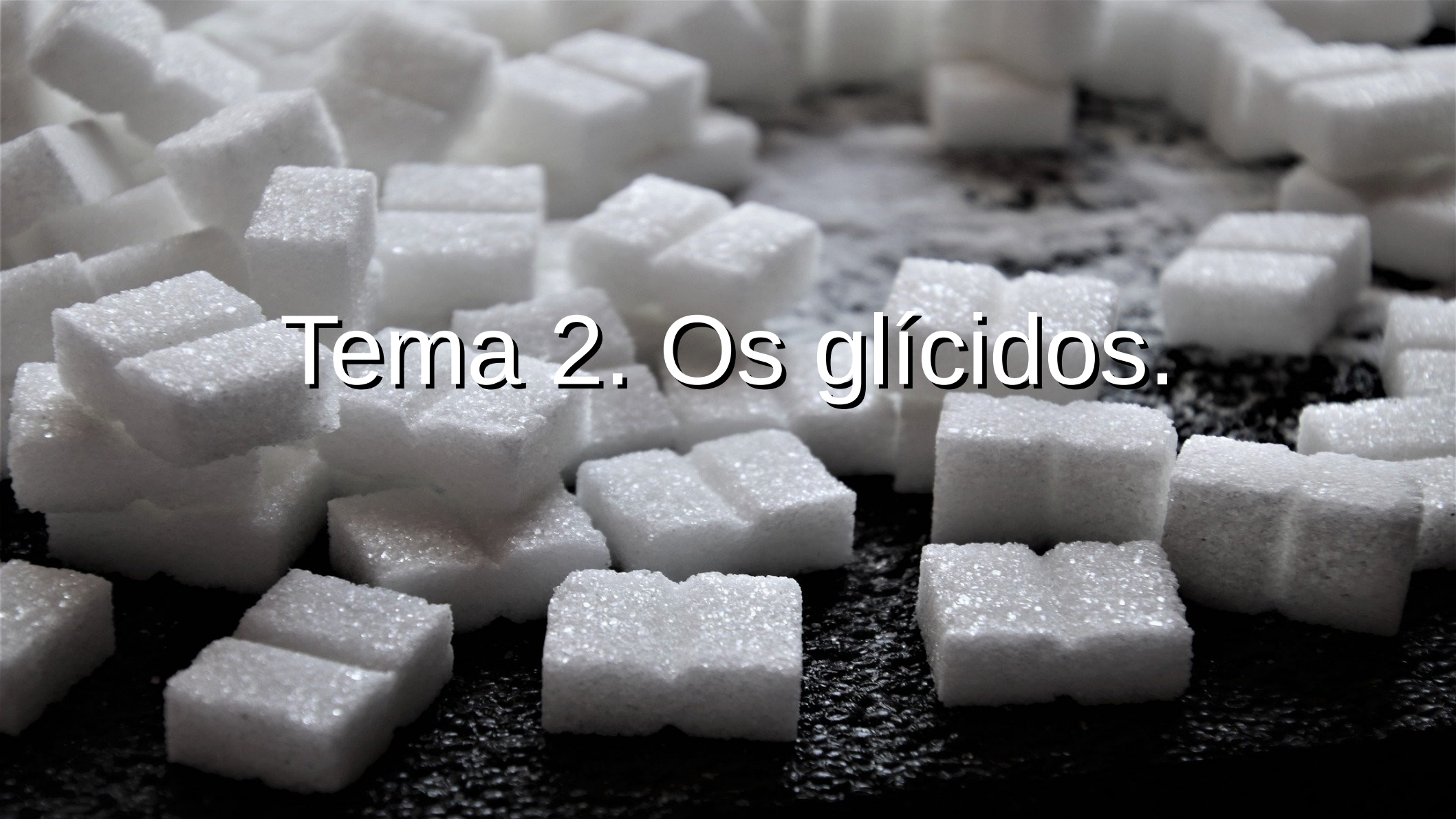


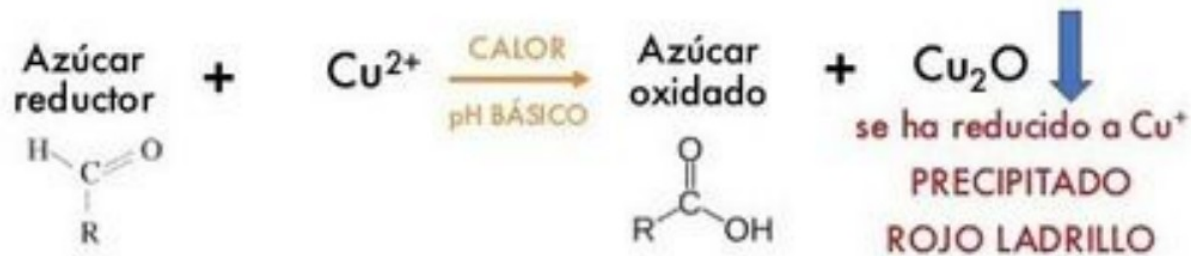
A close-up photograph of numerous white sugar cubes scattered across a dark, textured surface. The lighting is dramatic, highlighting the edges and facets of the cubes, while the background is blurred. The text 'Tema 2. Os glícidos.' is overlaid in the center in a white serif font.

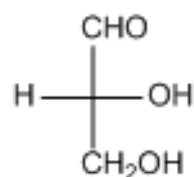
Tema 2. Os glícidos.



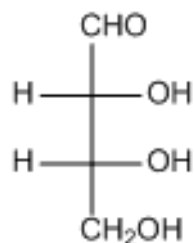
Monosacáridos

Detección de azúcares reductores (REACCIÓN DE FEHLING)

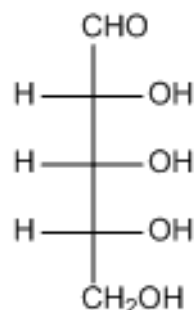




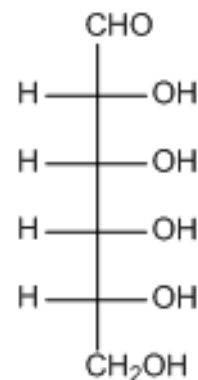
Aldosa de 3 carbonos
(aldotriosa)



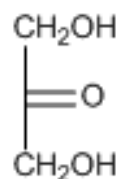
Aldosa de 4 carbonos
(aldotetrosa)



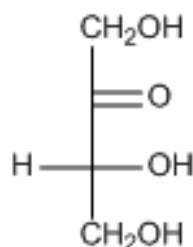
Aldosa de 5 carbonos
(aldopentosa)



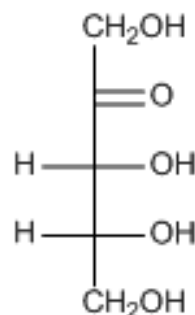
Aldosa de 6 carbonos
(Aldohexosa)



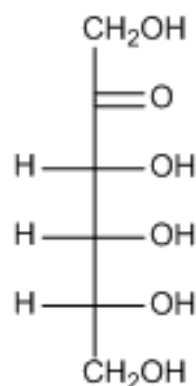
Cetosa de 3 carbonos
(cetotriosa)



Cetosa de 4 carbonos
(cetotetrosa)



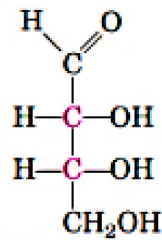
Cetosa de 5 carbonos
(cetopentosa)



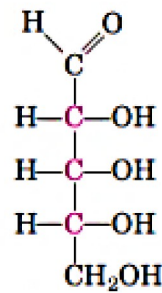
Cetosa de 6 carbonos
(cetohehexosa)

$$\begin{array}{c} \text{H} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ | \\ \text{H}-\text{C}-\text{OH} \\ | \\ \text{CH}_2\text{OH} \end{array}$$

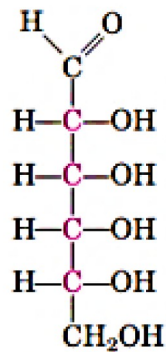
Quatro carbonos

O=C[C@@H](O)[C@H](O)CO

Cinco carbonos


$$\begin{array}{c} \text{H} \quad \text{O} \\ \diagdown \quad \diagup \\ \text{C} \\ | \\ \text{HO}-\text{C}-\text{H} \\ | \\ \text{H}-\text{C}-\text{OH} \\ | \\ \text{H}-\text{C}-\text{OH} \\ | \\ \text{CH}_2\text{OH} \end{array}$$
$$\begin{array}{c}
 \text{H} \quad \text{O} \\
 \diagdown \quad \diagup \\
 \text{C} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{CH}_2\text{OH}
 \end{array}$$
$$\begin{array}{c}
 \text{H} \quad \text{O} \\
 \diagdown \quad \diagup \\
 \text{C} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{CH}_2\text{OH}
 \end{array}$$

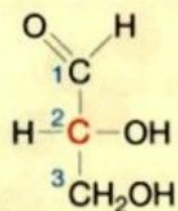
Seis carbonos

O=C[C@@H](O)[C@H](O)[C@H](O)CO
$$\begin{array}{c}
 \text{H} \quad \text{O} \\
 \diagdown \quad \diagup \\
 \text{C} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{CH}_2\text{OH}
 \end{array}$$
$$\begin{array}{c}
 \text{H} \quad \text{O} \\
 \diagdown \quad \diagup \\
 \text{C} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{CH}_2\text{OH}
 \end{array}$$
O=C[C@@H](O)[C@H](O)[C@@H](O)COO=C[C@@H](O)[C@H](O)[C@@H](O)CO
$$\begin{array}{c}
 \text{H} \quad \text{O} \\
 \diagdown \quad \diagup \\
 \text{C} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{HO}-\text{C}-\text{H} \\
 | \\
 \text{H}-\text{C}-\text{OH} \\
 | \\
 \text{CH}_2\text{OH}
 \end{array}$$
O=C[C@@H](O)[C@H](O)[C@@H](O)CO

D-Talose

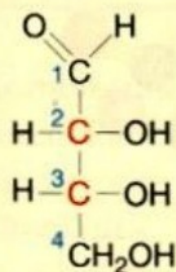
ALDOSAS

Triosas



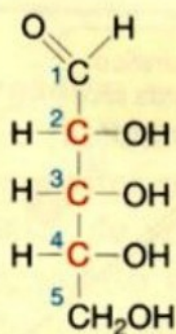
D-gliceraldehído

Tetrosas

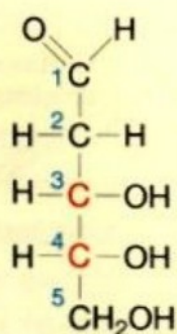


D-eritrosa

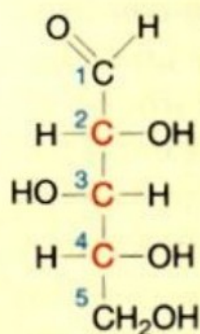
Pentosas



D-ribosa

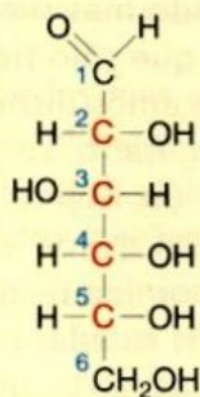


desoxirribosa

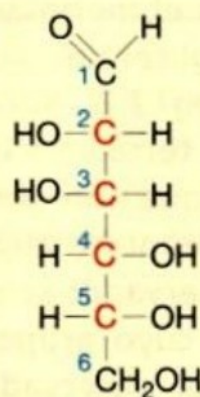


D-xilosa

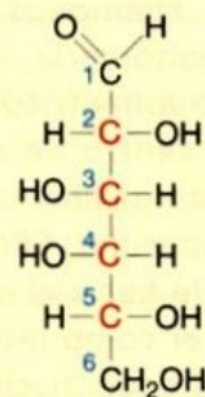
Hexosas



D-glucosa

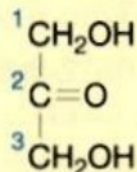


D-manosa

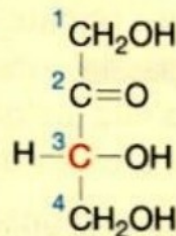


D-galactosa

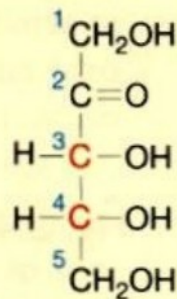
CETOSAS



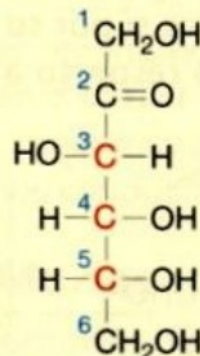
Dihidroxicetona



D-eritrolusa



D-ribulosa



D-fructosa

Isomería de función

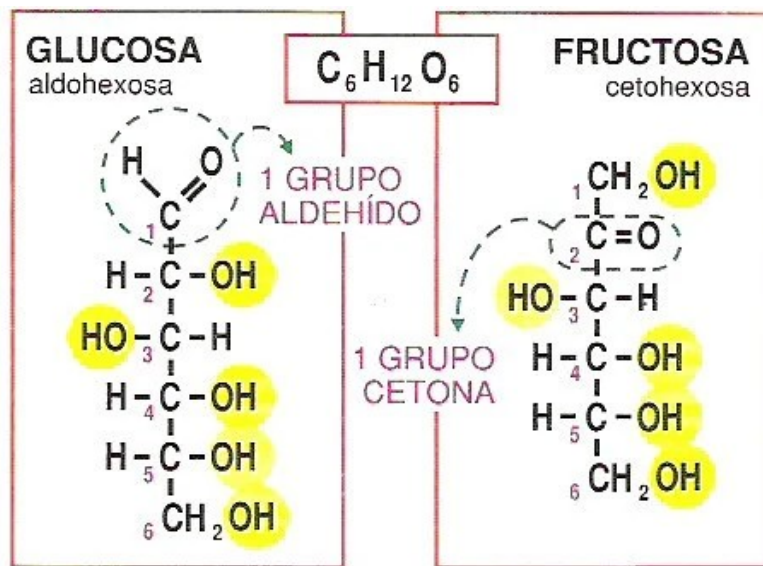
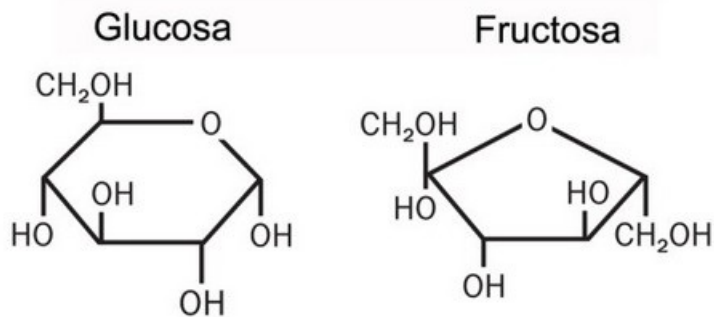
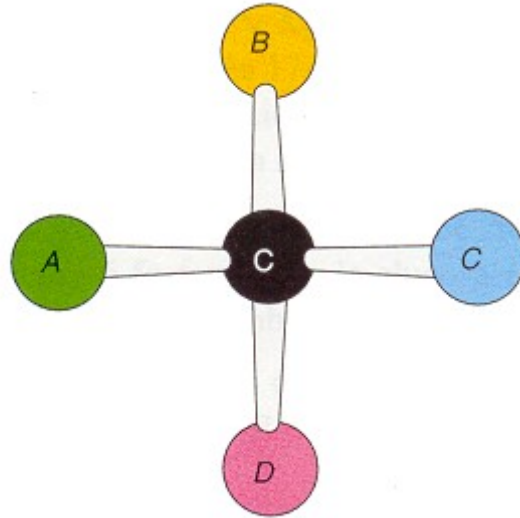
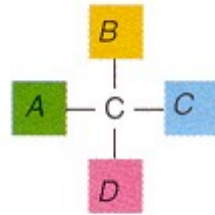


Figura 7.1



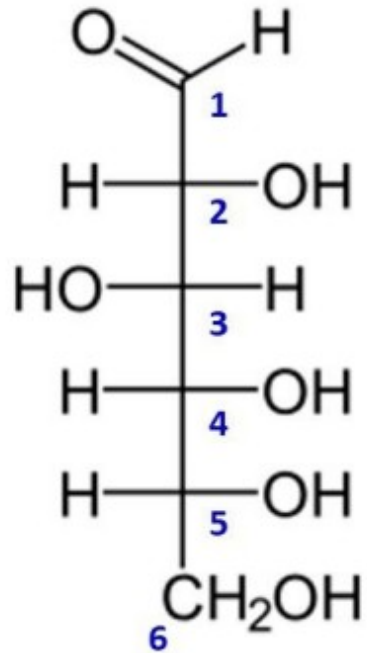
Proyección de
Haworth

Proyección de
Fischer

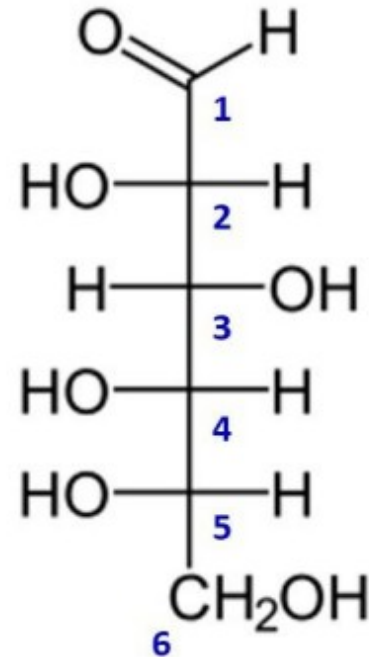


Estereoisomeria:
serie D e serie L.

Enantiómeros
D e L da
glicosa.



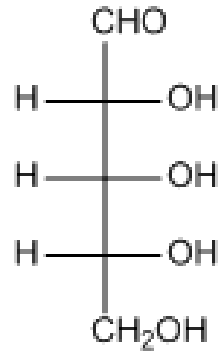
D-Glucosa



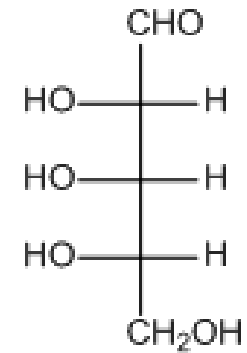
L-Glucosa

Estereoisomeria:
serie D e serie L.

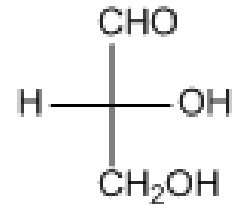
Enantiómeros
D e L da ribosa
e do
gliceraldeído.



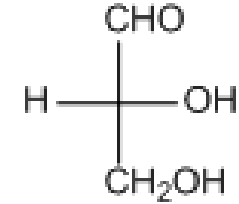
D-Ribosa



L-Ribosa



D-Gliceraldeído



L-Gliceraldeído

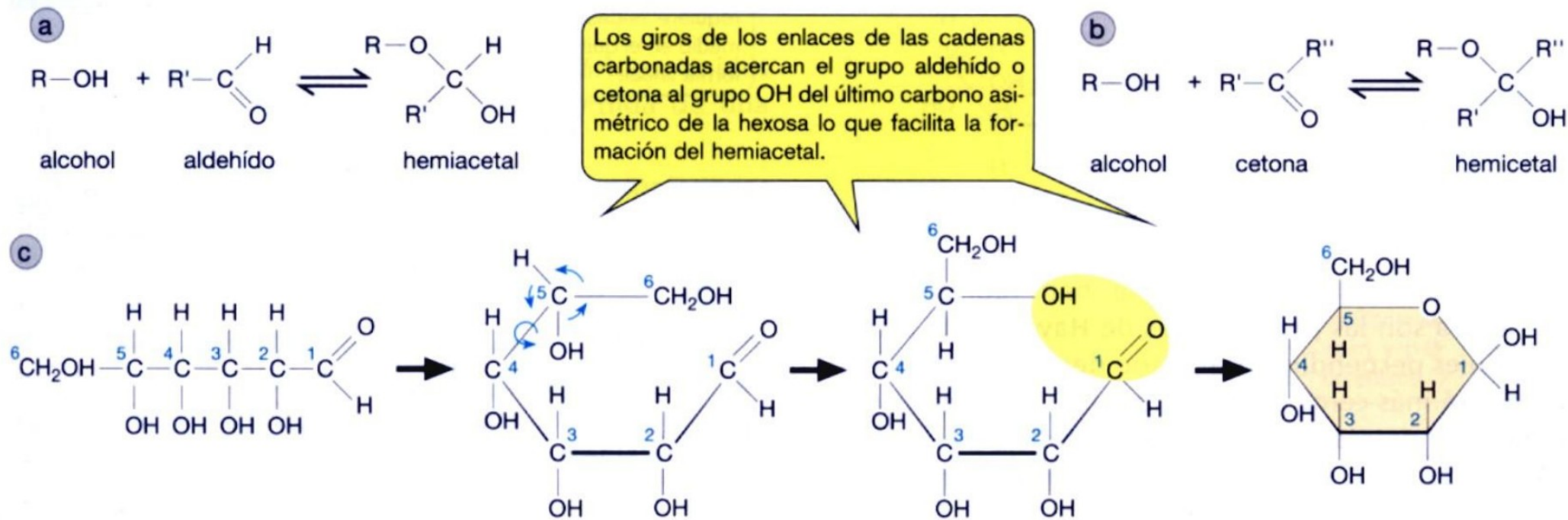
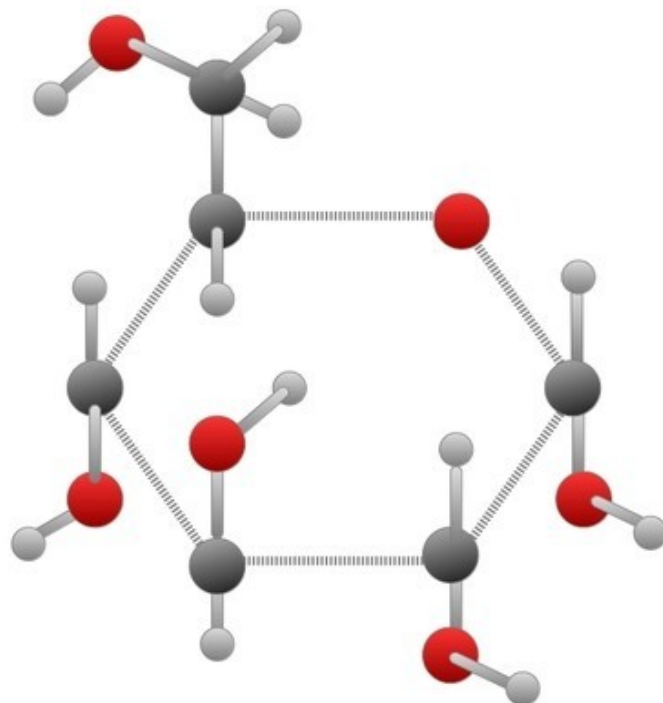
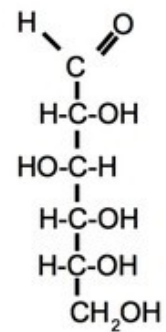


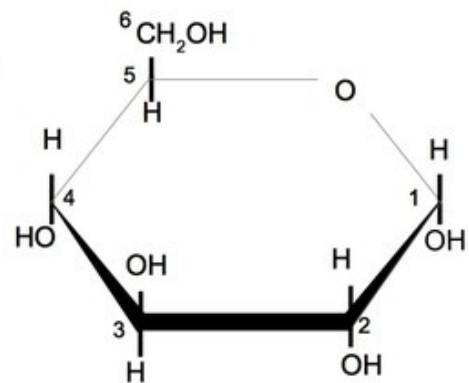
Fig. 2.6. a) Formación de un hemiacetal; b) Formación de un hemiacetal; c) Ciclado de una hexosa por formación de un hemiacetal.



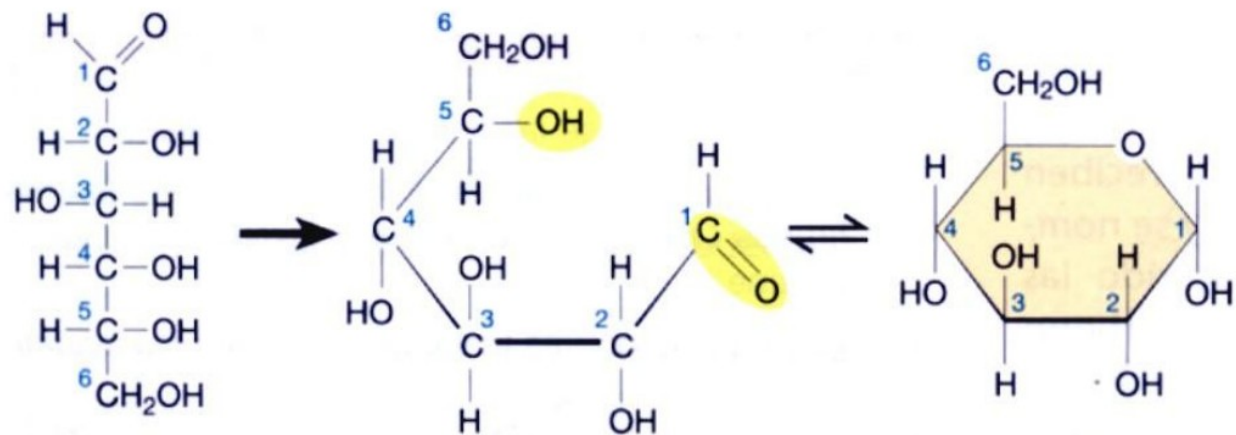
● Carbono
 ● Oxígeno
 ● Hidrógeno



Forma lineal



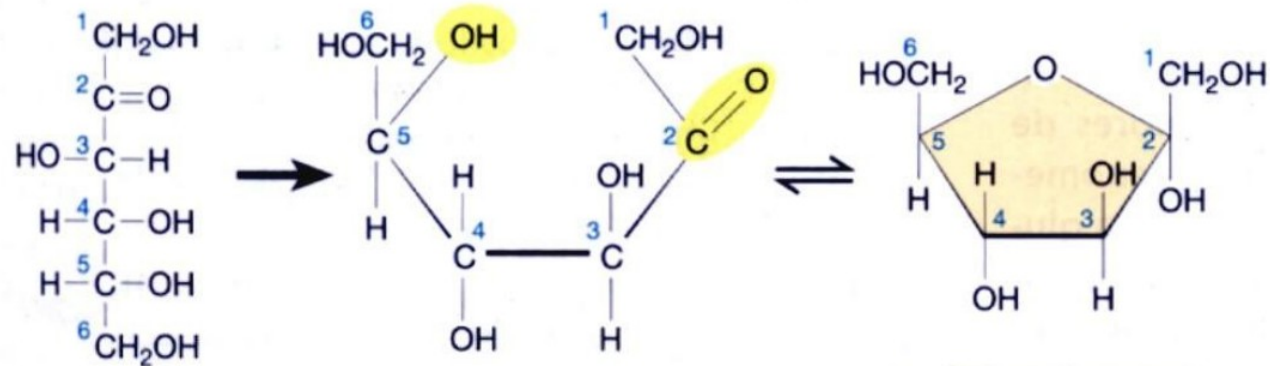
Forma cíclica



D-glucosa (forma lineal)

α -D-glucopiranososa
(proyección de Haworth)

La formación de un hemiacetal en la glucosa produce una molécula hexagonal (piranososa).

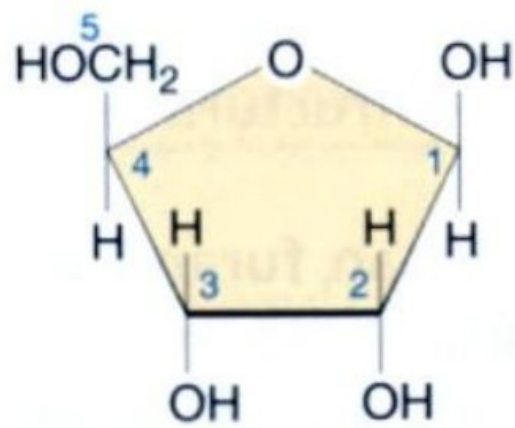


D-fructosa (forma lineal)

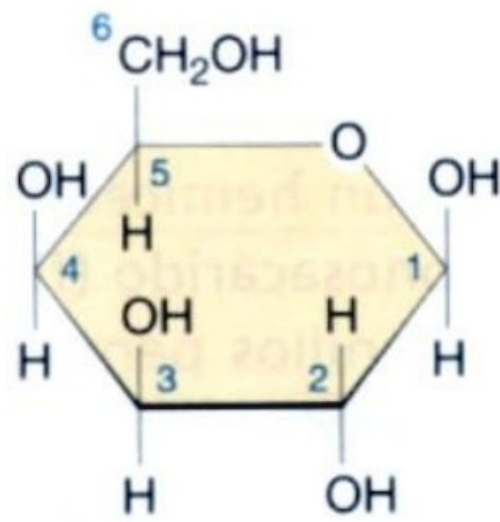
α -D-fructofuranosa
(proyección de Haworth)

La formación de un hemiacetal en la fructosa produce una molécula pentagonal (furanosa).

◀Fig. 2.7. Formación de las estructuras cíclicas de la D-glucosa y la D-fructosa.



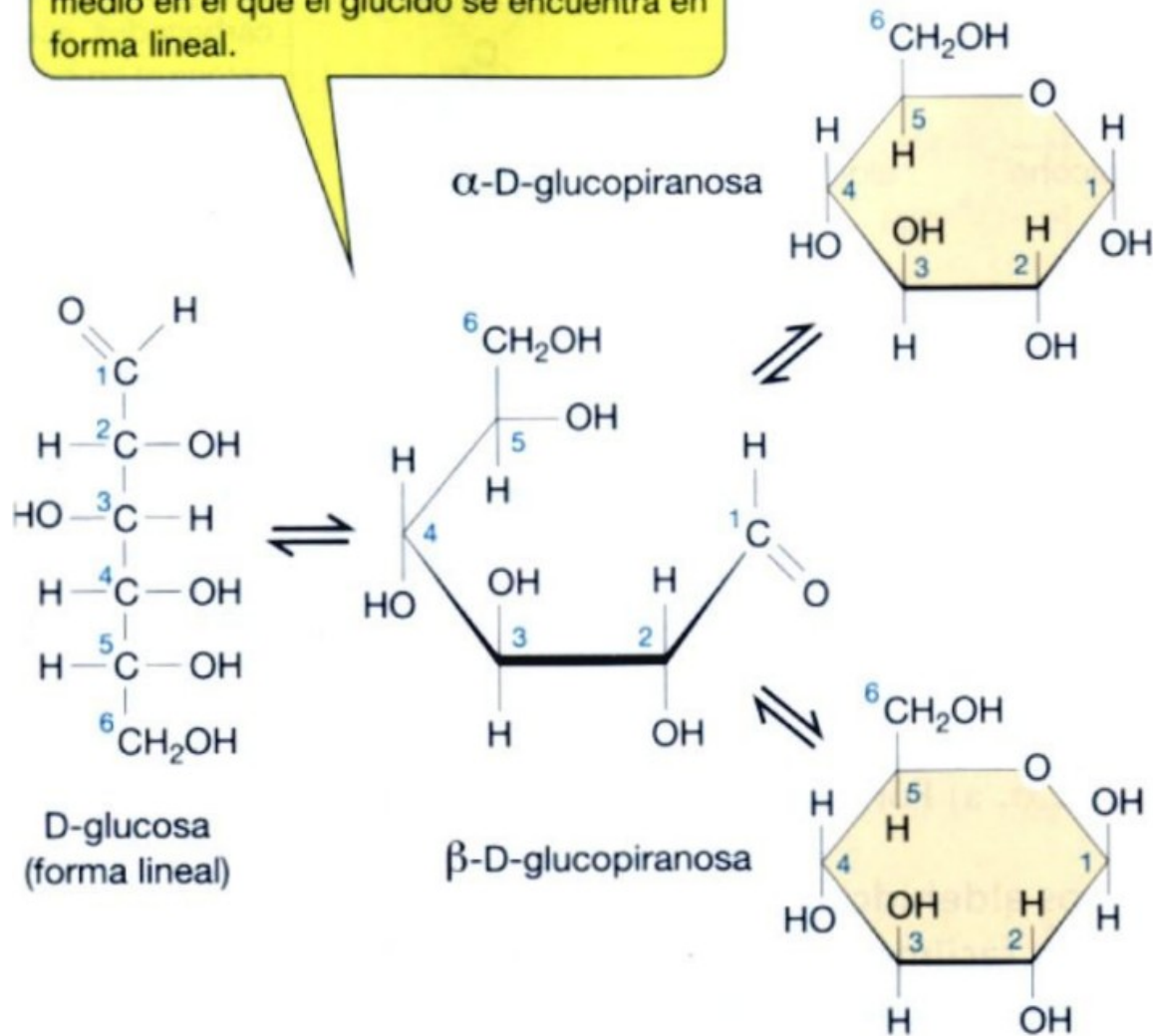
D-ribosa

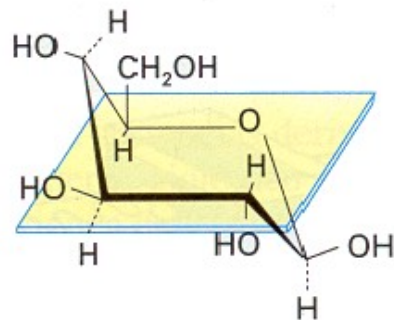


D-galactosa

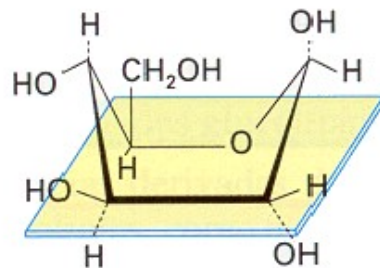
Fig. 2.9. Estructura cíclica de la D-ribosa y la D-galactosa.

La interconversión $\alpha \rightleftharpoons \beta$ (mutarrotación) requiere necesariamente un estadio intermedio en el que el glúcido se encuentra en forma lineal.





β -D - glucopiranososa en estructura de "silla".
 Los extremos de la molécula están en diferentes lados respecto al plano.



β -D - glucopiranososa en estructura de "nave".
 Los extremos de la molécula están en el mismo lado. Es una forma muy inestable.



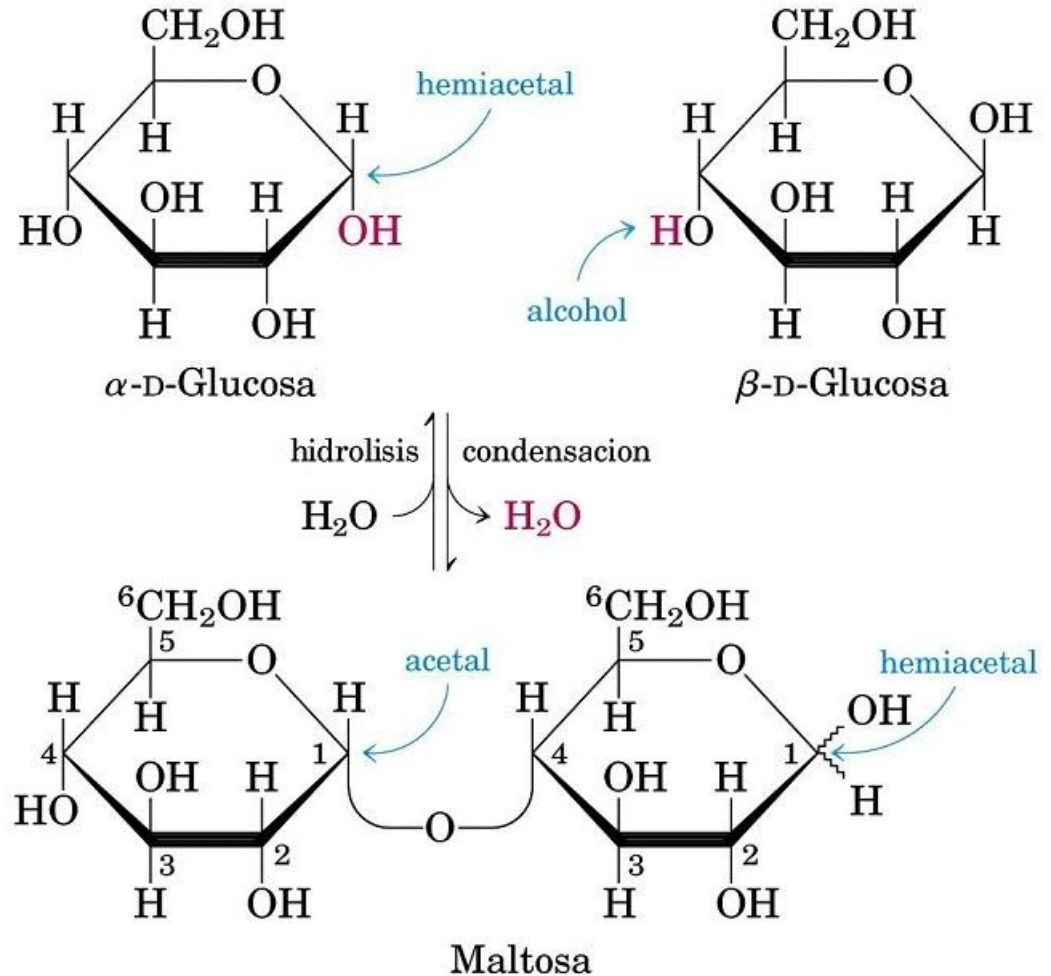
Disacáridos

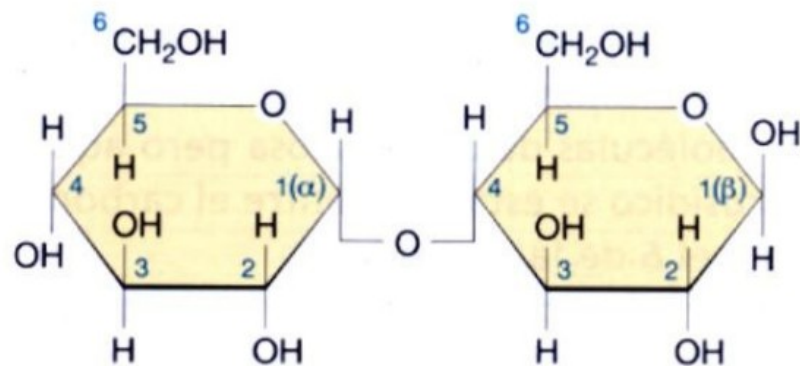
Enlace o-glicosídico

Enlace o-glicosídico α (1 $\rightarrow$ 4):

- A primeira osa é anómero α .
- Os -OH que participan no enlace son o 1 da primeira osa e o 4 da segunda.

Enlace monocarbonílico





Glucosa

Glucosa

Maltosa (forma β): α -D- glucopiranosil- (1 $\rightarrow$ 4)- β -D- glucopiranososa

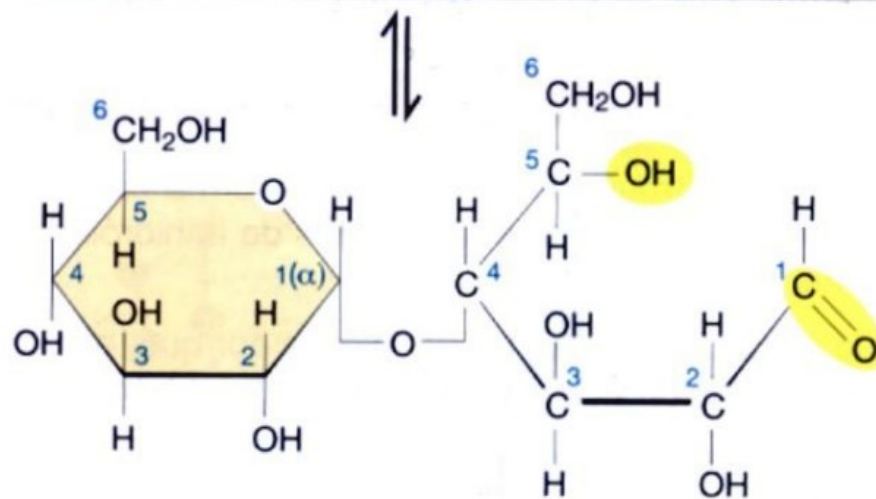


Fig. 3.2. Maltosa. a) Estructura; b) Proceso de mutarrotación.

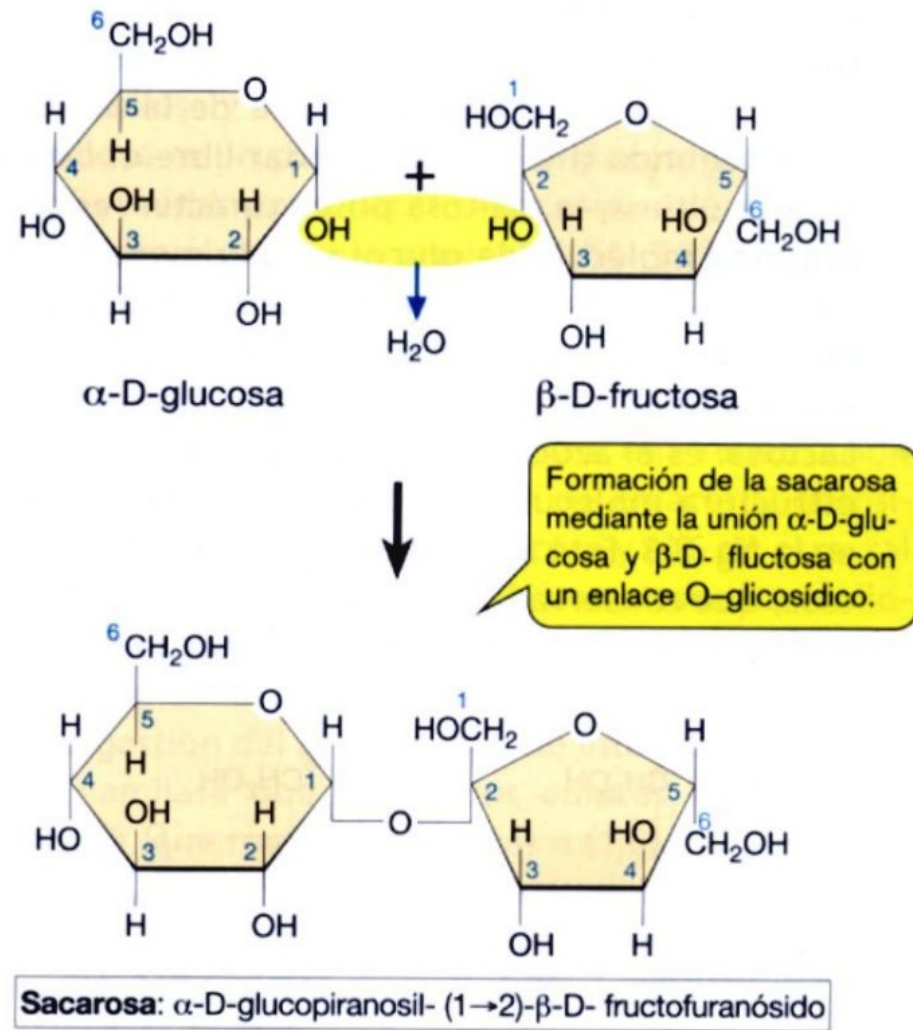
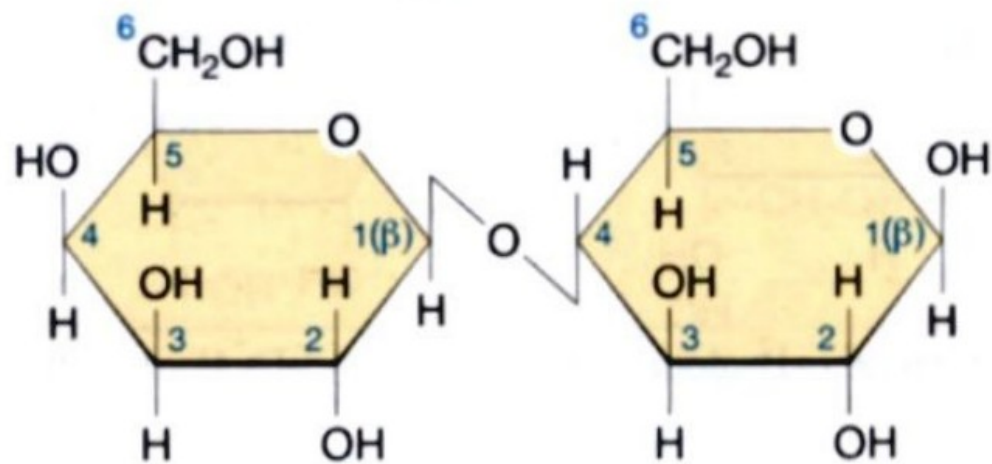


Fig. 3.1. Formación de la sacarosa mediante la unión de la α -D-glucosa y la β -D-fructosa

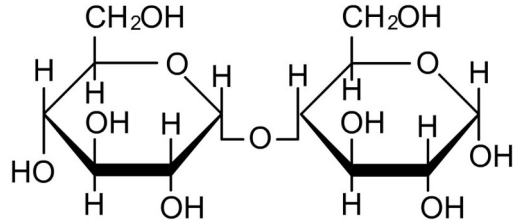


Galactosa

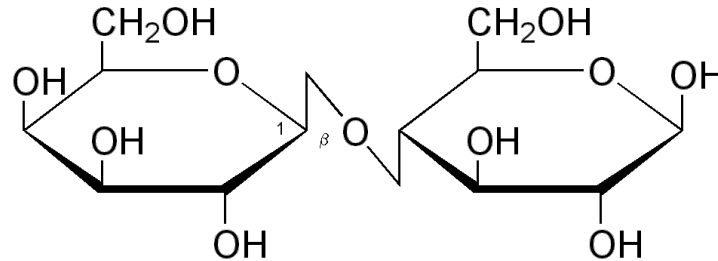
Glucosa

Lactosa (forma β): β - D- galactopiranosil- (1 $\rightarrow$ 4)- β - D- glucopiranososa

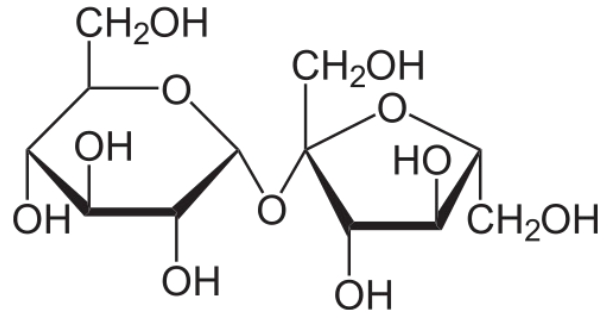
Maltosa: α – D – glicopiranosil – (1 $\rightarrow$ 4)- α/β – D – glicopiranososa

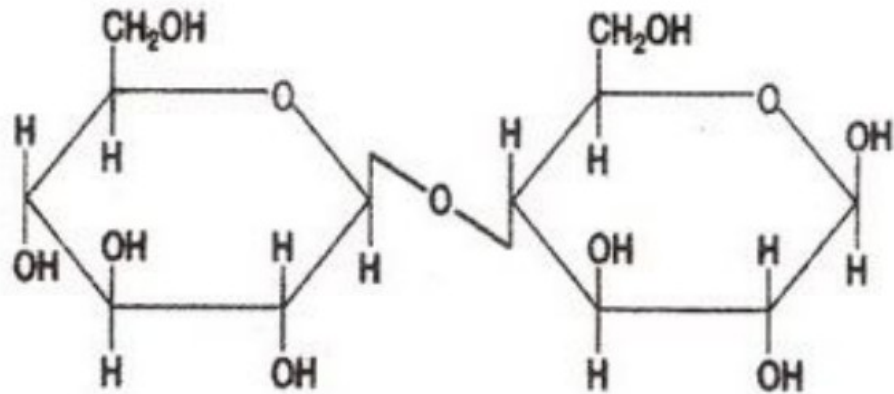
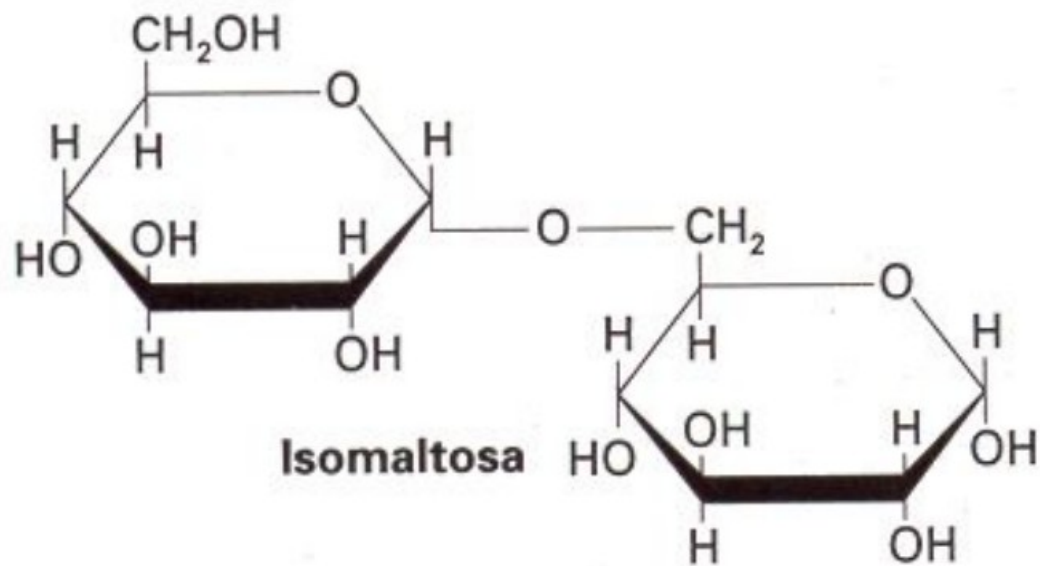


Lactosa: β – D – galactopiranosil – (1 $\rightarrow$ 4)- α/β – D – glicopiranososa



Sacarosa: α – D – glicopiranosil – (1 $\rightarrow$ 2)- β – D – fructofuranósido

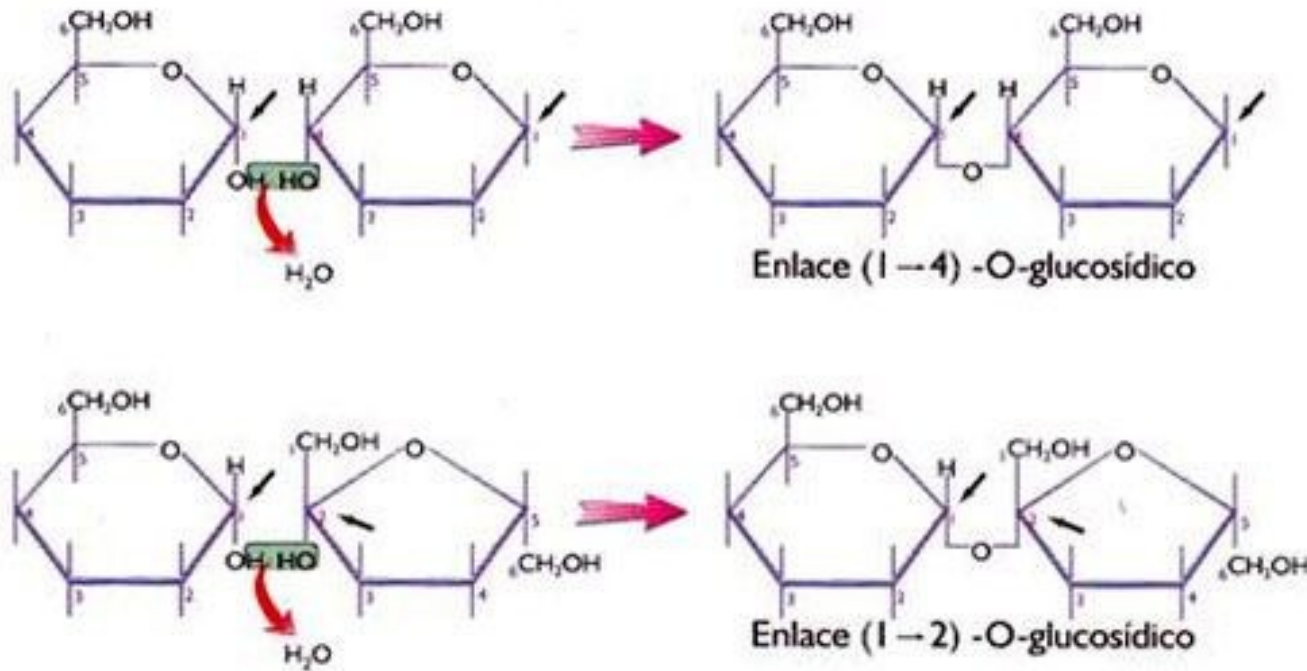




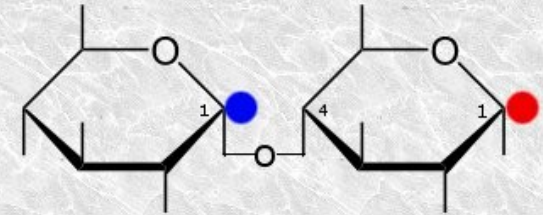
CELOBIOSA

O-β-D-glucopiranosil (1- 4)- β-D-glucopiranososa

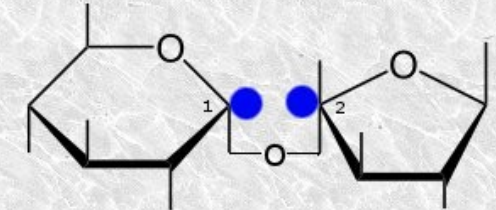
Os disacáridos con enlace dicarbonílico perden o poder reductor e polo tanto dan negativo na proba de Fehling (o reactivo non cambia de azul a vermello)



Disacárido reductor (enlace monocarbonílico)



Disacárido no reductor (enlace dicarbonílico)



- Poder reductor ON
- Poder reductor OFF

Figura 7.10



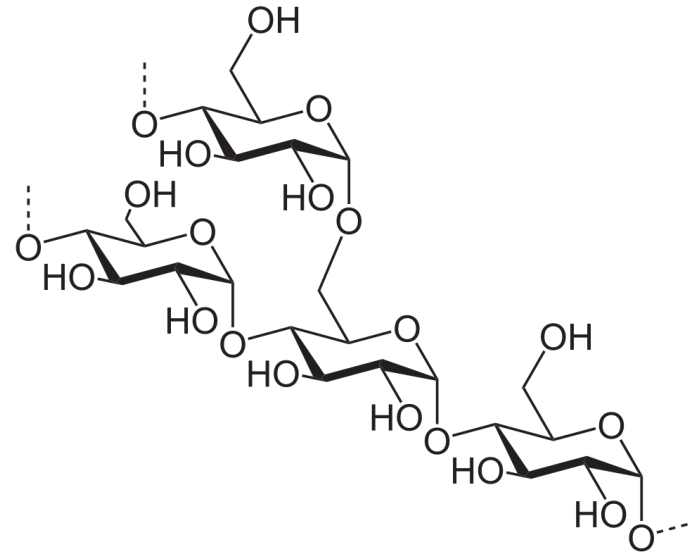
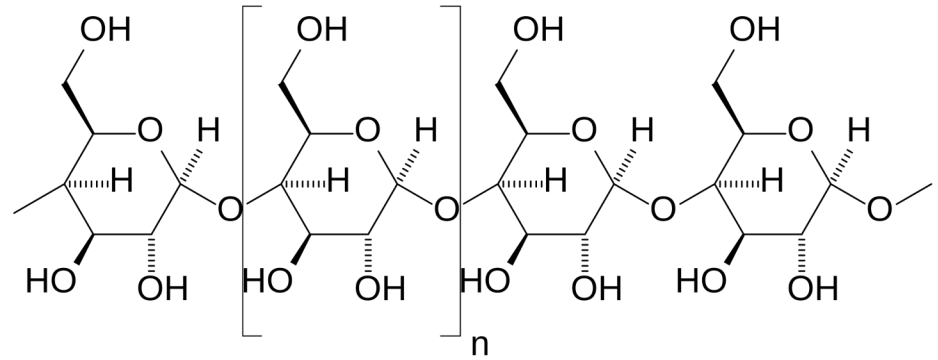
Polisacáridos

- Moitas osas (de decenas a milleiros) unidos por enlaces O-glicosídicos
- Sen poder redutor
- Non son solubles en auga, sen sabor e non son cristalizables
- Poden ser
 - Homopolisacáridos
 - Heteropolisacáridos

Homopolisacáridos de reserva

Amidón:

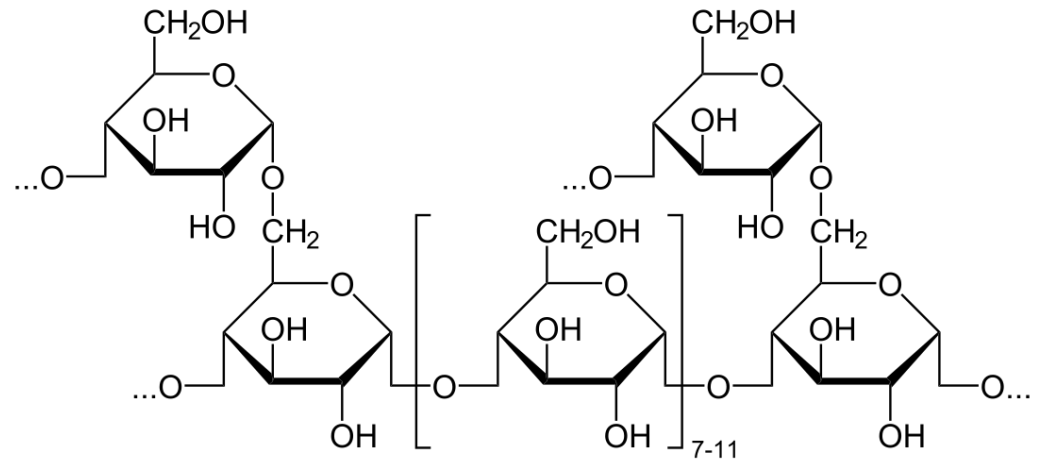
- Reserva das células vexetais.
- Formado por **α -D-glicosas**:
 - **Amilosa**: $\alpha(1 \rightarrow 4)$
 - **Amilopectina**: con ramificacións $\alpha(1 \rightarrow 6)$



Homopolisacáridos de reserva

Glicógeno:

- Reserva das células animais.
- Formado por **α -D-glicosas** unidas mediante enlaces $\alpha(1 \rightarrow 4)$ con ramificações $\alpha(1 \rightarrow 6)$ cada 8-10 moléculas



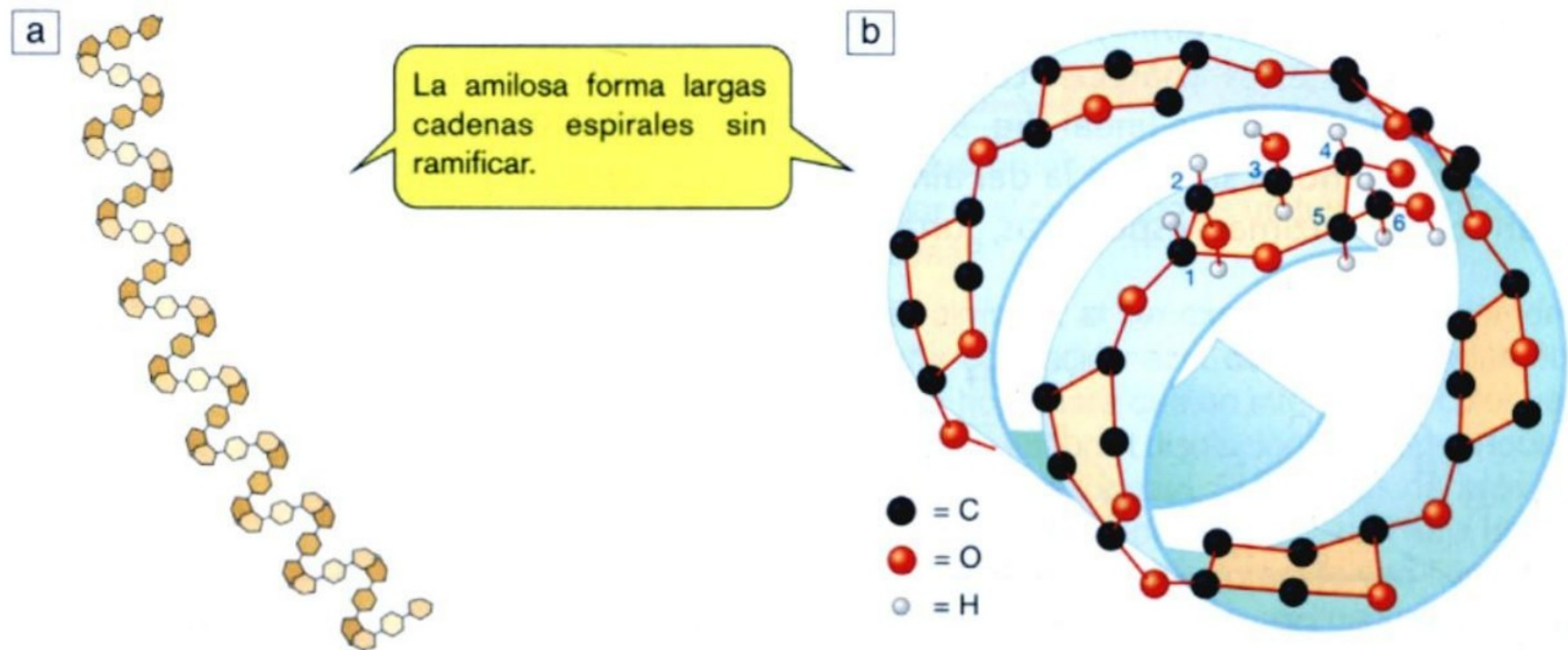
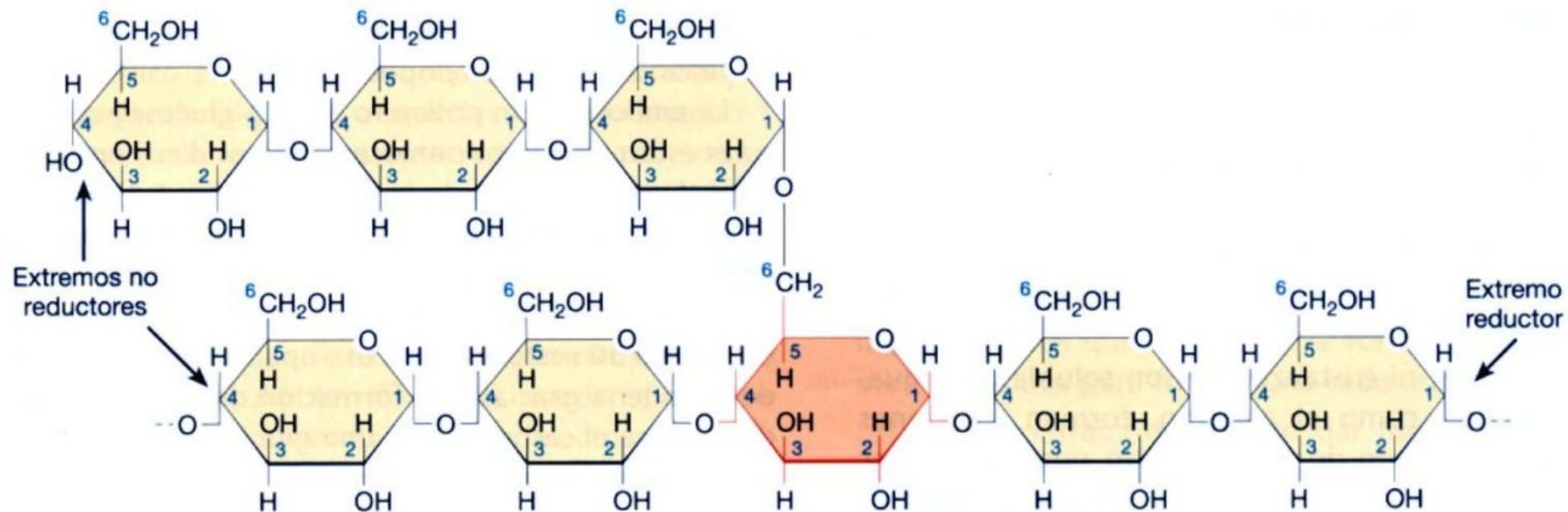


Fig. 5.2. Disposición helicoidal de la molécula de amilosa. a) aspecto general de la cadena; b) detalle de la misma.

a



Estructura ramificada común a la amilopectina y al glucógeno

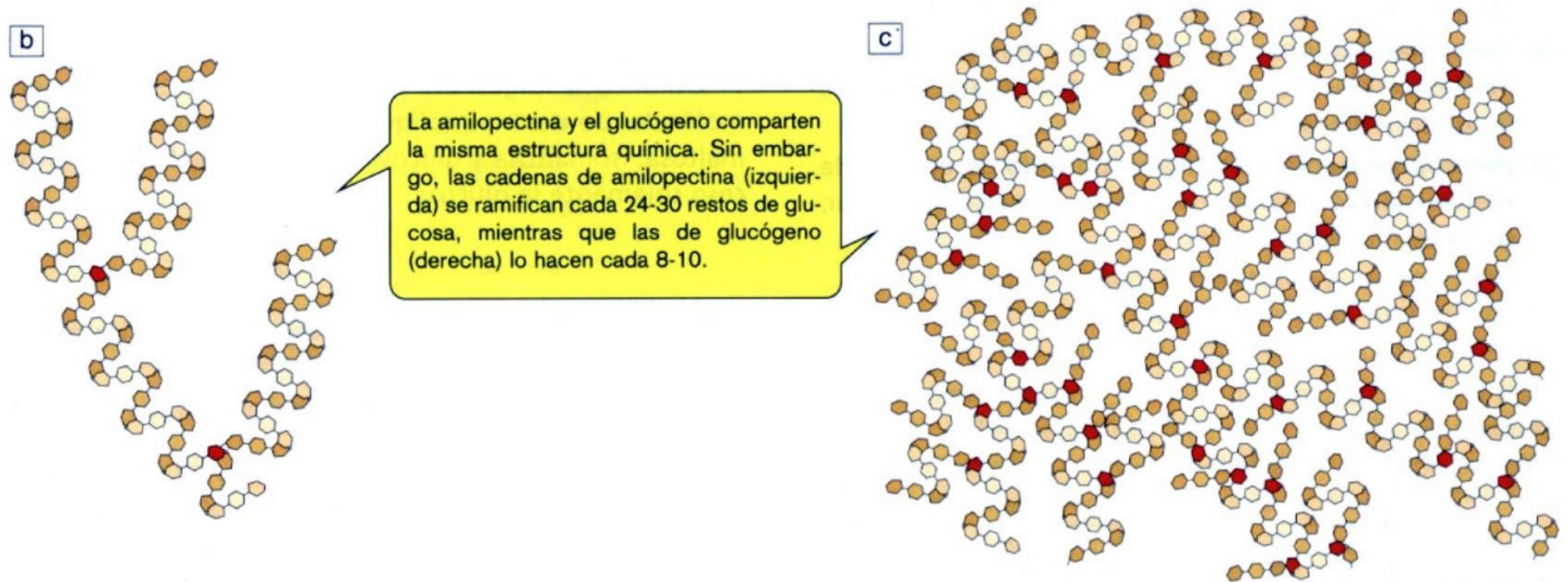
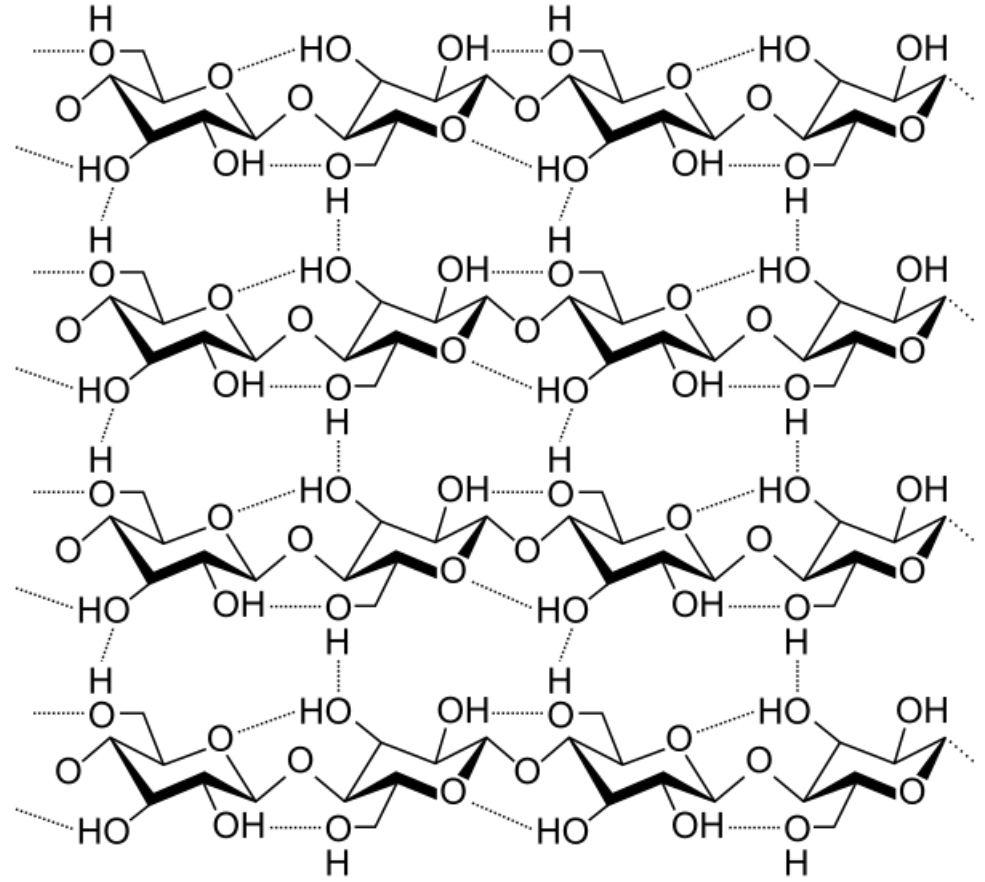


Fig. 5.4. a) Estructura ramificada común de la amilopectina y del glucógeno; b) cadena de amilopectina y c) cadena de glucógeno.

Homopolisacáridos estructurais

Celulosa:

- Paredes celulares vexetais.
- **β -D-glicosas** unidas por enlaces $\beta(1 \rightarrow 4)$ (celobiosas)
- Non temos ningún enzima que rompa o enlace $\beta(1 \rightarrow 4)$, por iso non a podemos dixerir.



Homopolisacáridos estruturais

Quitina:

- Exoesqueleto de artrópodos, parede celular de fungos.
- Cadeas lineais de N-acetil-D-glicosamina dispostas em paralelo. Enlaces $\beta(1\rightarrow4)$.

